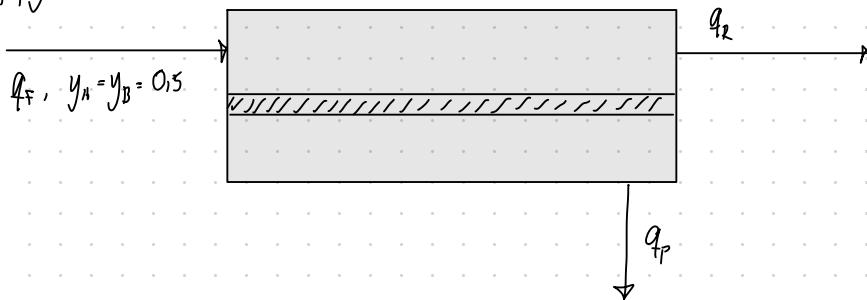


Oppgave 1



$$a) \quad y_{om} = \frac{y_F \left[1 + (\alpha^* - 1) \frac{P_P}{P_F} (1 - y_F) \right]}{\alpha^* (1 - y_F) + y_F}$$

$$= \frac{0.5 \left[1 + (30 - 1) \frac{1}{10} (1 - 0.5) \right]}{30(1 - 0.5) + 0.5} = 0.08$$

$$\Rightarrow \underline{y_B = 0.92}$$

$$b) \quad 0.08 \cdot 1.25 = 0.1$$

$$q_F y_A = q_R y_{RA} + q_P y_{PA}$$

$$2 \cdot 0.5 = q_R \cdot 0.1 + q_P y_{PA}$$

$$y_{PA} = \frac{1 - 0.1 \cdot 0.94}{1.06}$$

$$\underline{y_{PA} = 0.855}$$

$$y_{PB} = 0.145$$

$$q_{PB} = q_P \cdot y_{PB} = 1.06 \cdot 0.145$$

$$\Theta = \frac{q_D}{q_F} \Rightarrow q_P = \Theta q_F = 0.53 \cdot 2 = 1.06$$

$$\Rightarrow q_R = 0.94$$

$$\% \text{ loss} = \frac{1,06 \cdot 0,145}{0,5 \cdot 2} = \underline{\underline{15,4\%}}$$

c) Se LF

$$A = 357 \text{ m}^2$$

d) - Cascade

- Pressure ratio

- Increase selectivity

2

Mass balance top:

$$D \cdot x_d + L_n x_n = V_{n+1} y_{n+1}$$

$$y_{n+1} = \frac{L_n}{V_{n+1}} x_n + \frac{D x_d}{V_{n+1}}$$

$$D = V - L$$

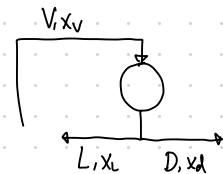
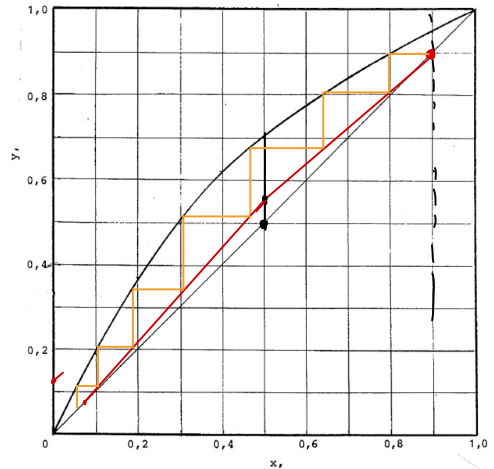
$$\Rightarrow y_{n+1} = \frac{L}{V} x_n + \frac{V}{V} x_d - \frac{L}{V} x_d$$

$$y_{n+1} = \frac{L}{V} (x_n - x_d) + x_d, \quad x_d = 0,9 \quad \frac{L}{V} = 0,86$$

$$y_{n+1} = 0,86(0 - 0,9) + 0,9 = 0,126$$

$$y_{n+1} = 0,86(0,9 - 0,9) + 0,9 = 0,9$$

Vet ikke x_B , og må finne med prøv og feil $x_B \approx 0,07$



Feed is liquid

$\Rightarrow$ q-line = vertical

X_B for oss finne D og B

$$F = D + B \Rightarrow B = F - D$$

$$F x_F = D y_d + B x_B$$

$$10 \cdot 0,5 = D \cdot 0,9 + (10 - D) \cdot 0,07$$

$$\rightarrow D = \frac{10(0,5 - 0,07)}{0,9 - 0,07} = \underline{5,14}$$

$$B = \underline{4,82}$$

ii) Moving it changes the composition $X_B \approx 0,6$

b)

MV CV	L	VB/QB	D/B	LCw(Qc)
L/V	•			
y _d		•		
MD				•
MB			•	
P			•	

Disturbances F, x_F

Se lf.

3 a) $V_1 = 3000 \text{ mol/s}$ $y_1 = 0,002$

$L_2 = 90.000 \text{ mol/s}$ $x_2 = 0$

Veldig fortynnet $\Rightarrow$ Anter $V_1 = V_2 = V$, $L_1 = L_2 = L$

Ved likevekt: $P_{s02} = H x \Rightarrow y_P = H x \Rightarrow y = \frac{H}{P} x$, $\frac{H}{P} = m$

1 bar, 25°C

$$m = \frac{46}{1} = 46$$

$$A = \frac{L/V}{m} = \frac{90\,000/3000}{46} = 0,652$$

Like vekt på bunnen:

$$X_1 = x^*$$

$$x^* = \frac{y_1}{m} = \frac{0,002}{46} = 4,35 \cdot 10^{-5}$$

Finer y_2 ved massebalanse av SO_2

$$I_{\text{nn}} = U_1$$

$$V y_1 = V y_2 + L x_1$$

$$y_2 = y_1 - \frac{L}{V} x_1$$

$$y_2 = y_1 - \frac{L}{V} \cdot \frac{y_1}{m}$$

$$y_2 = y_1 \left(1 - \frac{L}{V m}\right) = y_1 (1 - A)$$

$$y_2 = 0,002(1 - 0,652)$$

$$y_2 = 6,96 \cdot 10^{-4} \gg \text{ønsket}$$

Må ikke vædestrommen.

b) For at $y_1' = y_2$

$$\Rightarrow y_2' = y_1'(1-A)$$

$$\Rightarrow y_2' = 0,000242 \text{ fortsett for stort}$$

c) Alle der A, og gtr greit resultat

Se LF

d) g) Assumes dilute mixtures with m and $\frac{1}{V} = \text{constant}$

b) "Imaginary" vapor compositions in equilibrium with x_1 and x_2

$$y_1^* = m x_1$$

$$y_2^* = m x_2$$

c) Se LF

4) $T = 28^\circ\text{C}$, $H = 0,3$, $\rho_b = 712,8 \text{ kg/m}^3$

$$C_p = 0,01 \frac{\text{kg H}_2\text{O}}{\text{kg perst}}$$

$$V = 4052 \frac{\text{kg}}{\text{m}^2 \cdot \text{h}}, \quad C_0 = 926 \cdot 10^{-6} \frac{\text{kg H}_2\text{O}}{\text{kg H}_2}$$

a) Gideler ikke

$$b) \tau_b = \int_0^{t_b} \left(1 - \frac{c}{c_0}\right) dt = t_b$$

$$t_b \approx 9,58 \text{ h}$$

$$c) H_{unb} = \left(1 - \frac{t_b}{t_s}\right) H_1 = \left(1 - \frac{9,58}{12,6}\right) 0,3 = 0,072 \text{ m}$$

$$\tau_r = t_b + \frac{t_d - t_b}{2} = 11,19 = t_s$$

$$\text{Frem til } t_b: \quad \frac{t_b}{t_s} = 85,6\%$$

d) Etter 12,6 h, er absorbert H_2O :

$$\begin{aligned} m_{\text{H}_2\text{O}} &= 4052 \frac{\text{kg H}_2\text{O}}{\text{m}^2 \cdot \text{h}} \cdot 926 \cdot 10^{-6} \frac{\text{kg H}_2\text{O}}{\text{kg N}_2} \\ &= 375 \frac{\text{kg H}_2\text{O}}{\text{m}^2 \cdot \text{h}} \cdot 9,58 \text{ h} \\ &= 35,95 \text{ kg H}_2\text{O}/\text{m}^2 \\ &= \frac{35,95 \text{ kg H}_2\text{O}/\text{m}^2}{(0,3 - 0,075) \text{ m}} = \underline{\underline{160 \text{ kg H}_2\text{O}/\text{m}^3}} \end{aligned}$$